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# Crystal and Molecular Structure of $\boldsymbol{\beta}$-5-Isopropyl-2'-deoxyuridine 

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(Received 30 January 1979; accepted 20 March 1979)


#### Abstract

$\mathrm{C}_{12} \mathrm{H}_{18} \mathrm{~N}_{2} \mathrm{O}_{5}$ is monoclinic, space group $P 2_{1}$, with $a=$ 7.999 (1), $b=16.266$ (6), $c=4.855$ (3) $\AA, \beta=$ 94.34 ( 8$)^{\circ}, Z=2$. The final $R=0.061$ for 992 counter reflexions. $\mathrm{C}(8)$ of the isopropyl moiety lies in the best plane of the pyrimidine ring while $\mathrm{C}(7)$ and $\mathrm{C}(9)$ are displaced by 0.22 and $1.59 \AA$. The glycosidic torsion angle $\mathrm{O}\left(1^{\prime}\right)-\mathrm{C}\left(1^{\prime}\right)-\mathrm{N}(1)-\mathrm{C}(6)$ is $64.4^{\circ}$ (anti conformation). The sugar ring has the usual $\mathrm{C}\left(2^{\prime}\right)$-endo conformation. The arrangement about $\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)$ is such that $\mathrm{O}\left(5^{\prime}\right)$ is oriented gauche-gauche $\left(g^{+}\right)$with respect to $\mathrm{O}\left(1^{\prime}\right)$ and $\mathrm{C}\left(3^{\prime}\right)$.


## Introduction

The polymerization of a series of 5 -alkyl-substituted $2^{\prime}$ deoxyuridine triphosphates by Escherichia coli DNA polymerase-I enzyme has recently been studied in our laboratory (Sági, Szabolcs, Szemzö \& Ötvös, 1977). A decreased reaction rate of the title compound with respect to that of the other derivatives has been thought to be correlated with some steric effect of the isopropyl moiety. In order to shed light on this question, the conformations of the title compound and $\beta$-5-ethyl- 2 '-

0567-7408/79/071626-04\$01.00
deoxyuridine have been determined in the crystalline state. Our structure determinations form part of the series of X-ray analyses of 5 -substituted $2^{\prime}$-deoxyuridines (Barr, Hamor \& Walker, 1978; and references herein) studied as potential antiviral and antitumor agents (Torrence, Spencer, Bobst, Descamps \& De Clercq, 1978).

## Experimental

The title compound and other 5 -alkyl-2'-deoxyuridine derivatives studied in enzymatic reactions (Sági et al., 1977) were synthesized in our laboratory (Szabolcs, Sági \& Ötvös, 1975; Ötvös, Szabolcs, Sági \& Szemzö, 1975). Cell dimensions were determined from Weissenberg and precession photographs. Intensities were measured on a Stoe-Güttinger two-circle diffractometer with Ni -filtered Cu Ka radiation (Kálmán, Simon, Schawartz \& Horvath, 1974). 1002 reflexions were scanned, of which 992 with $I-2 \cdot 5 \sigma(I) \geq 0$ were used in the analysis. No absorption correction was made.

## Crystal data

$\mathrm{C}_{12} \mathrm{H}_{18} \mathrm{~N}_{2} \mathrm{O}_{5}, \quad M_{r}=270 \cdot 29$, monoclinic, $a=$ 7.999 (1), $b=16.266$ (6), $c=4.855$ (3) $\AA, \beta=$ © 1979 International Union of Crystallography
$94.34(8)^{\circ}, U=629.9 \AA^{3}, Z=2, D_{c}=1.425 \mathrm{Mg} \mathrm{m}^{-3}$, $F(000)=288$; systematic absences: $0 k 0, k$ odd; space group $P 2_{1}, \mathrm{Cu} K \alpha$ radiation, $\lambda=1 \cdot 5418 \AA$.

## Structure determination

The structure was solved by direct methods with 140 normalized structure factors having $E \geq 1.40$. An $E$ map computed from a phase set with the second best consistency gave the positions of 17 of the 19 nonhydrogen atoms. The missing two atoms were located from a difference synthesis. Blocked-full-matrix leastsquares refinement of positional and vibrational parameters (anisotropic for non-hydrogen and an overall isotropic temperature factor $\bar{u}=0.057 \AA^{2}$ for H atoms) converged to a final $R$ of 0.061 for 992 reflexions $\left\{R_{w}\right.$ $=\sum w^{1 / 2} \Delta / \sum w^{1 / 2}\left|F_{o}\right|=0.072$ with the final weighting scheme $\left.w=1 /\left[\sigma^{2}\left(F_{o}\right)+0 \cdot 00642 F_{o}^{2}\right]\right\}$. The H atoms belonging to OH moieties were located in a difference map and were kept fixed at the last stage of refinement. The positions of the other 16 H atoms were generated from assumed geometries with $\mathrm{C}-\mathrm{H}$ and $\mathrm{N}-\mathrm{H}$ distances constrained to 1.08 and $0.94 \AA$. A bondedH -atom scattering factor was employed (Stewart, Davidson \& Simpson, 1965) with complex neutral scattering factors for the remaining atoms (Cromer \& Mann, 1968; Cromer \& Liberman, 1970). The calculations were performed with SHELX (Sheldrick, 1976). The atomic coordinates are given in Table 1.*

[^0]
## Discussion

The shape of the molecule is illustrated in Fig. 1, which also shows the numbering of the atoms. Bond lengths and angles are in Table 2 and the results of least-squares-plane calculations in Table 3. Selected torsion angles together with relevant data from the literature are summarized in Table 4.

## Stereochemistry and conformation of the molecule

Bond lengths and angles agree with corresponding values in 5-vinyl-2'-deoxyuridine (Hamor, O'Leary \& Walker, 1978), 5-ethynyl-2'-deoxyuridine (Barr, Hamor \& Walker, 1978), sodium uridine-5'-O-methyl-phosphate-methanol (Hoogendorp \& Romers, 1978) and related compounds. Accordingly, the effect of the vinyl and ethynyl moieties on the electronic structure of the pyrimidine ring is shown by the bond distances and angles at $C(5)$.
(a) The deoxyribose residue. The mean-plane calculations (Table 3) and the corresponding torsion



Fig. 1. A stereoscopic drawing of the title compound showing the atomic numbering.

Table 1. Fractional atomic coordinates $\left(\times 10^{4}\right)$ and e.s.d.'s
Owing to constrained refinements, the e.s.d.'s of the H atoms are identical to those given for the corresponding non- H atoms.

|  | $x$ | $y$ | $z$ |  | $x$ | $y$ |  |
| :--- | ---: | ---: | ---: | :--- | ---: | ---: | ---: |
|  |  |  |  |  |  |  |  |
| $\mathrm{~N}(1)$ | $8465(3)$ | $5129(2)$ | $2199(4)$ | $\mathrm{H}(\mathrm{N} 3)$ | 8231 | 7062 | 1703 |
| $\mathrm{C}(2)$ | $8855(4)$ | $5903(2)$ | $1250(6)$ | $\mathrm{H}(\mathrm{C} 6)$ | 6808 | 4392 | 4420 |
| $\mathrm{~N}(3)$ | $7975(3)$ | $6530(2)$ | $2300(5)$ | $\mathrm{H}(\mathrm{C} 81)$ | 5593 | 4481 | 8594 |
| $\mathrm{C}(4)$ | $6730(3)$ | $6471(2)$ | $4159(5)$ | $\mathrm{H}(\mathrm{C} 82)$ | 4280 | 4245 | 5618 |
| $\mathrm{C}(5)$ | $6203(3)$ | $5635(2)$ | $4741(5)$ | $\mathrm{H}(\mathrm{C} 83)$ | 3409 | 4631 | 8587 |
| $\mathrm{C}(6)$ | $7120(4)$ | $5012(2)$ | $3844(6)$ | $\mathrm{H}(\mathrm{C} 91)$ | 3247 | 6394 | 3821 |
| $\mathrm{C}(7)$ | $4673(3)$ | $5523(2)$ | $6318(6)$ | $\mathrm{H}(\mathrm{C} 92)$ | 2036 | 5747 | 5800 |
| $\mathrm{C}(8)$ | $4474(4)$ | $4659(4)$ | $7351(6)$ | $\mathrm{H}(\mathrm{C} 93)$ | 2907 | 5361 | 2830 |
| $\mathrm{C}(9)$ | $3103(4)$ | $5774(3)$ | $4565(7)$ | $\mathrm{H}(\mathrm{C} 7)$ | 4847 | 5919 | 8101 |
| $\mathrm{O}(2)$ | $9903(4)$ | $6025(2)$ | $-425(6)$ | $\mathrm{H}\left(\mathrm{C} 1^{\prime}\right)$ | 10455 | 4682 | 402 |
| $\mathrm{O}(4)$ | $6141(3)$ | $7105(2)$ | $5047(5)$ | $\mathrm{H}(\mathrm{C} 21)$ | 9301 | 3856 | 5383 |
| $\mathrm{C}\left(1^{\prime}\right)$ | $9401(3)$ | 4427 | $1375(5)$ | $\mathrm{H}(\mathrm{C} 22)$ | 11348 | 4067 | 4460 |
| $\mathrm{O}\left(1^{\prime}\right)$ | $8315(3)$ | $3924(2)$ | $-348(4)$ | $\mathrm{H}\left(\mathrm{C} 3^{\prime}\right)$ | 10155 | 2540 | 3778 |
| $\mathrm{C}\left(2^{\prime}\right)$ | $10109(4)$ | $3869(2)$ | $3700(6)$ | $\mathrm{H}\left(\mathrm{C} 4^{\prime}\right)$ | 8972 | 2746 | -1657 |
| $\mathrm{C}\left(3^{\prime}\right)$ | $10168(4)$ | $3043(3)$ | $2326(5)$ | $\mathrm{H}(\mathrm{C} 51)$ | 7334 | 1998 | 1265 |
| $\mathrm{C}\left(4^{\prime}\right)$ | $8661(4)$ | $3075(2)$ | $167(6)$ | $\mathrm{H}(\mathrm{C} 52)$ | 6063 | 2762 | -406 |
| $\mathrm{C}\left(5^{\prime}\right)$ | $7105(4)$ | $2651(3)$ | $1096(6)$ | $\mathrm{H}\left(\mathrm{O} 5^{\prime}\right)$ | 5806 | 2655 | 3982 |
| $\mathrm{O}\left(5^{\prime}\right)$ | $6722(3)$ | $2958(2)$ | $3663(5)$ | $\mathrm{H}\left(\mathrm{O} 3^{\prime}\right)$ | 12558 | 2813 | 1951 |
| $\mathrm{O}\left(3^{\prime}\right)$ | $11649(3)$ | $2974(2)$ | $877(5)$ |  |  |  |  |

angles (Table 4) indicate that the sugar ring has the $S$ type $C\left(2^{\prime}\right)$-endo conformation with $C\left(2^{\prime}\right)$ as the flap of an (approximate) envelope; this can also be seen from the lowest asymmetry parameter $\Delta C_{s^{\prime}}^{\mathrm{C}\left(2^{\prime}\right)}=3.4^{\circ}$

Table 2. Bond distances $(\AA)$ and angles $\left(^{\circ}\right)$

| $\mathrm{N}(1)-\mathrm{C}\left(1^{\prime}\right) \quad 1.4$ | 1.438 (4) | $\mathrm{C}(7)-\mathrm{C}(9) \quad 1.5$ | 1.519 (4) |
| :---: | :---: | :---: | :---: |
| $\mathrm{N}(1)-\mathrm{C}(2) \quad 1.38$ | 1.384 (5) | $\mathrm{O}\left(1^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right) \quad 1.4$ | 1.426 (5) |
| $\mathrm{N}(1)-\mathrm{C}(6) \quad 1.401$ | 1.401 (4) | $\mathrm{C}\left(1^{\prime}\right)-\mathrm{O}\left(1^{\prime}\right) \quad 1.42$ | 1.420 (3) |
| $\mathrm{C}(2)-\mathrm{N}(3) \quad 1.3$ | 1.361 (5) | $\mathrm{C}\left(1^{\prime}\right)-\mathrm{C}\left(2^{\prime}\right) \quad 1.5$ | 1.524 (4) |
| $\mathrm{C}(2)-\mathrm{O}(2) \quad 1.2$ | 1.227 (4) | $\mathrm{C}\left(2^{\prime}\right)-\mathrm{C}\left(3^{\prime}\right) \quad 1.50$ | 1.502 (6) |
| $\mathrm{N}(3)-\mathrm{C}(4) \quad 1.3$ | 1.397 (4) | $\mathrm{C}\left(3^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right) \quad 1.5$ | 1.537 (4) |
| C(4)-C(5) 1.4 | 1.458 (6) | $\mathrm{C}\left(3^{\prime}\right)-\mathrm{O}\left(3^{\prime}\right) \quad 1.427$ | 1.427 (4) |
| $\mathrm{C}(4)-\mathrm{O}(4) \quad 1.2$ | 1.226 (5) | $\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right) \quad 1.5$ | 1.521 (5) |
| $\mathrm{C}(5)-\mathrm{C}(6) \quad 1.3$ | 1.342 (5) | $\mathrm{C}\left(5^{\prime}\right)-\mathrm{O}\left(5^{\prime}\right) \quad 1.3$ | 1.398 (4) |
| $\mathrm{C}(5)-\mathrm{C}(7) \quad 1.503$ | 1.503 (4) | $\mathrm{O}\left(5^{\prime}\right)-\mathrm{H}\left(\mathrm{O5}^{\prime}\right) \quad 0.9$ | 0.906 (4) |
| $\mathrm{C}(7)-\mathrm{C}(8) \quad 1.5$ | 1.504 (7) | $\mathrm{O}\left(3^{\prime}\right)-\mathrm{H}\left(\mathrm{O3}^{\prime}\right) \quad 0.901(3)$ |  |
| $\mathrm{C}(2)-\mathrm{N}(1)-\mathrm{C}(6)$ | 121.0 (3) | $\mathrm{C}(8)-\mathrm{C}(7)-\mathrm{C}(9)$ | 109.6 (3) |
| $\mathrm{C}(2)-\mathrm{N}(1)-\mathrm{C}\left(1^{\prime}\right)$ | 119.6 (3) | $\mathrm{N}(1)-\mathrm{C}(6)-\mathrm{C}(5)$ | 123.0 (4) |
| $\mathrm{C}\left(1^{\prime}\right)-\mathrm{N}(1)-\mathrm{C}(6)$ | 119.3 (3) | $\mathrm{N}(1)-\mathrm{C}\left(1^{\prime}\right)-\mathrm{O}\left(1^{\prime}\right)$ | 108.3 (3) |
| $\mathrm{N}(1)-\mathrm{C}(2)-\mathrm{N}(3)$ | 114.9 (3) | $\mathrm{N}(1)-\mathrm{C}\left(1^{\prime}\right)-\mathrm{C}\left(2^{\prime}\right)$ | 116.1 (3) |
| $\mathrm{N}(1)-\mathrm{C}(2)-\mathrm{O}(2)$ | 123.4 (4) | $\mathrm{O}\left(1^{\prime}\right)-\mathrm{C}\left(1^{\prime}\right)-\mathrm{C}\left(2^{\prime}\right)$ | $105 \cdot 6$ (2) |
| $\mathrm{O}(2)-\mathrm{C}(2)-\mathrm{N}(3)$ | 121.7 (4) | $\mathrm{C}\left(1^{\prime}\right)-\mathrm{C}\left(2^{\prime}\right)-\mathrm{C}\left(3^{\prime}\right)$ | $103 \cdot 1$ (2) |
| $\mathrm{C}(2)-\mathrm{N}(3)-\mathrm{C}(4)$ | 127.2 (4) | $\mathrm{C}\left(2^{\prime}\right)-\mathrm{C}\left(3^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)$ | 102.9 (3) |
| $\mathrm{N}(3)-\mathrm{C}(4)-\mathrm{O}(4)$ | 118.7 (4) | $\mathrm{C}\left(2^{\prime}\right)-\mathrm{C}\left(3^{\prime}\right)-\mathrm{O}\left(3^{\prime}\right)$ | $110 \cdot 1$ (3) |
| $\mathrm{N}(3)-\mathrm{C}(4)-\mathrm{C}(5)$ | 114.9 (4) | $\mathrm{O}\left(3^{\prime}\right)-\mathrm{C}\left(3^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)$ | 107.7 (2) |
| $\mathrm{O}(4)-\mathrm{C}(4)-\mathrm{C}(5)$ | 126.3 (3) | $\mathrm{O}\left(1^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(3^{\prime}\right)$ | 106.5 (3) |
| $\mathrm{C}(4)-\mathrm{C}(5)-\mathrm{C}(6)$ | 117.9 (3) | $\mathrm{O}\left(1^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)$ | 109.8 (3) |
| $\mathrm{C}(4)-\mathrm{C}(5)-\mathrm{C}(7)$ | 118.0 (3) | $\mathrm{C}\left(3^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)$ | 113.6 (3) |
| $\mathrm{C}(6)-\mathrm{C}(5)-\mathrm{C}(7)$ | $124 \cdot 1$ (4) | $\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)-\mathrm{O}\left(5^{\prime}\right)$ | $110 \cdot 0$ (4) |
| $\mathrm{C}(5)-\mathrm{C}(7)-\mathrm{C}(8)$ | 113.2 (3) | $\mathrm{C}\left(1^{\prime}\right)-\mathrm{O}\left(1^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)$ | $110 \cdot 6$ (2) |
| $\mathrm{C}(5)-\mathrm{C}(7)-\mathrm{C}(9)$ | 110.6 (3) |  |  |

Table 3. Mean-plane calculations
Deviations are in $\AA$; atoms whose deviations are marked by an asterisk define the plane. $X, Y, Z$ refer to the orthogonal axial system; weighted r.m.s.d.'s are $0.043 \AA$ for $a$ and $0.013 \AA$ for $d$.

|  | Plane $a$ | $b$ | $c$ | $d$ |
| :--- | ---: | ---: | ---: | ---: |
| $\mathrm{~N}(1)$ | $0.050(3)^{*}$ | $0.005(9)^{*}$ | $0.007^{*}$ |  |
| $\mathrm{C}(2)$ | $-0.039(4)^{*}$ | $-0.014(8)^{*}$ | $-0.006^{*}$ |  |
| $\mathrm{~N}(3)$ | $-0.018(3)^{*}$ | $0.021(9)^{*}$ | $-0.002^{*}$ |  |
| $\mathrm{C}(4)$ | $0.062(3)^{*}$ | $-0.018(9)^{*}$ | $0.009^{*}$ |  |
| $\mathrm{C}(5)$ | $-0.054(4)^{*}$ | $0.009(8)^{*}$ | $-0.009^{*}$ |  |
| $\mathrm{C}(6)$ | $-0.001(4)^{*}$ | $-0.003(9)^{*}$ | $0.001^{*}$ |  |
| $\mathrm{O}(2)$ | $-0.140(4)$ | $0.009(9)$ | 0.010 |  |
| $\mathrm{O}(4)$ | $0.164(4)$ | $-0.072(8)$ | 0.027 |  |
| $\mathrm{C}(7)$ | $-0.218(3)$ | $0.002(10)$ | -0.025 |  |
| $\mathrm{C}(8)$ | $-0.015(5)$ | $-0.173(10)$ | 0.177 |  |
| $\mathrm{C}(9)$ | $-1.585(5)$ | - | - |  |
| $\mathrm{C}\left(1^{\prime}\right)$ | $0.131(3)$ | $0.076(9)$ | 0.027 | $0.010(3)^{*}$ |
| $\mathrm{C}\left(2^{\prime}\right)$ |  |  |  | $0.509(3)$ |
| $\mathrm{C}\left(3^{\prime}\right)$ |  |  |  | $-0.009(3)^{*}$ |
| $\mathrm{C}\left(4^{\prime}\right)$ |  |  |  | $0.016(4)^{*}$ |
| $\mathrm{O}\left(1^{\prime}\right)$ |  |  |  | $-0.016(3)^{*}$ |
| $\mathrm{C}\left(5^{\prime}\right)$ |  |  |  | $1.241(4)$ |

Plane $a: 0.6556 X+0.0566 Y+0.7530 Z=5.2666$.
Plane $b$ : 5-ethyl-2'-deoxyuridine (M. Czugler, A. Kálmán \& J. T. Sági, unpublished work).
Plane $c: 5$-vinyl-2'-deoxyuridine (Hamor et al., 1978).
Plane $d$ (deoxyribose ring):
$-0.6355 X-0.0220 Y+0.7718 Z=-4.8578$.
(Duax, Weeks \& Rohrer, 1976) and the pseudorotation phase angle of $169.9^{\circ}$ with the maximum amplitude of puckering $\varphi_{m}=33.9^{\circ}$ [if $\varphi_{o}$ is assigned to the rotation about $\mathrm{O}\left(1^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)$ (Altona, Geise \& Romers, 1968)]. The conformation about $\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)$ is gauche-gauche $\left(g^{+}\right)$(Sundaralingam, 1973), as revealed by the torsion angles $\mathrm{O}\left(1^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)-$ $O\left(5^{\prime}\right)=66.4^{\circ}$ and $C\left(3^{\prime}\right)-C\left(4^{\prime}\right)-C\left(5^{\prime}\right)-O\left(5^{\prime}\right)=$ $53.0^{\circ}$. A similar arrangement has been reported in sodium uridine-5'-O-methylphosphate-methanol (Hoogendorp \& Romers, 1978) and 5-iodo-2'-deoxyuridine (Camerman \& Trotter, 1965). The glycosidic torsion angle $\chi=64.4^{\circ}$ falls in the part of the anti range found most frequently for $\mathrm{C}\left(2^{\prime}\right)$-endo $\beta$ pyrimidine nucleosides. A similar $\chi\left(63.3^{\circ}\right)$ was observed in 5-iodo-2'-deoxyuridine (Camerman \& Trotter, 1965). In sodium uridine-5'-O-methyl-phosphate-methanol (Hoogendorp \& Romers, 1978) $\chi$ is $79.1^{\circ}$ while in 5-ethynyl-2'-deoxyuridine (Barr, Hamor \& Walker, 1978) it is only $18.6^{\circ}$. The angle between the best planes of the sugar ring (for four atoms) and the pyrimidine base (for six atoms) is $80 \cdot 6^{\circ}$.
(b) The pyrimidine base. The pyrimidine ring is less planar than in 5 -vinyl-2'-deoxyuridine (Hamor, O'Leary \& Walker, 1978) and 5-ethyl-2'-deoxyuridine (refinement in progress) (Table 3). $\mathrm{C}\left(1^{\prime}\right), \mathrm{O}(2), \mathrm{O}(4)$ and $C(7)$ are out of the best plane of the base by $0.13-$ $0.22 \AA$. Nevertheless, the synperiplanar conformation of $\mathrm{C}(7)-\mathrm{C}(8)$ about $\mathrm{C}(5)-\mathrm{C}(7)$ resembles those in 5 -vinyl- and 5-ethyl-2'-deoxyuridines. Accordingly, the other methyl group of the isopropyl moiety should protrude significantly from the least-squares plane of the base ( $1.59 \AA$ ). If this asymmetric conformation of the isopropyl moiety is sustained in solution, even to some extent, then the protruding methyl group can hinder the development of base-stacking of the natural DNA chain. The disturbing effect of such a protruding methyl group in the formation of a DNA chain has been modeled as shown in Fig. 2. This hypothesis based on the molecular conformation observed in the crystal-

Table 4. Selected torsion angles ( ${ }^{\circ}$ ) of the title compound (I, mean e.s.d. $0 \cdot 8^{\circ}$ ) compared with those of sodium uridine-5'-O-methylphosphate-methanol (II)
(Hoogendorp \& Romers, 1978).

|  |  | (I) | (II) |
| :--- | :--- | :---: | :---: |
| $\tau_{0}$ | $\mathrm{C}\left(4^{\prime}\right)-\mathrm{O}\left(1^{\prime}\right)-\mathrm{C}\left(1^{\prime}\right)-\mathrm{C}\left(2^{\prime}\right)$ | $-17 \cdot 8$ | $-19 \cdot 8$ |
| $\tau_{1}$ | $\mathrm{O}\left(1^{\prime}\right)-\mathrm{C}\left(1^{\prime}\right)-\mathrm{C}\left(2^{\prime}\right)-\mathrm{C}\left(3^{\prime}\right)$ | $31 \cdot 5$ | $34 \cdot 3$ |
| $\tau_{2}$ | $\mathrm{C}\left(1^{\prime}\right)-\mathrm{C}\left(2^{\prime}\right)-\mathrm{C}\left(3^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)$ | $-32 \cdot 2$ | -34.9 |
| $\tau_{3}$ | $\mathrm{C}\left(2^{\prime}\right)-\mathrm{C}\left(3^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)-\mathrm{O}\left(1^{\prime}\right)$ | $22 \cdot 6$ | 24.6 |
| $\tau_{4}$ | $\mathrm{C}\left(3^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)-\mathrm{O}\left(1^{\prime}\right)-\mathrm{C}\left(1^{\prime}\right)$ | $-3 \cdot 0$ | -3.0 |
| $\psi^{\prime}$ | $\mathrm{O}\left(3^{\prime}\right)-\mathrm{C}\left(3^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)$ | $145 \cdot 3$ | $148 \cdot 2$ |
| $\psi$ | $\mathrm{C}\left(3^{\prime}\right)-\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)-\mathrm{O}\left(5^{\prime}\right)$ | $53 \cdot 0$ | $52 \cdot 8$ |
|  | $\mathrm{C}\left(4^{\prime}\right)-\mathrm{C}\left(5^{\prime}\right)-\mathrm{O}\left(5^{\prime}\right)-X$ | $-179 \cdot 0$ | $173 \cdot 4$ |
|  | $\mathrm{O}\left(1^{\prime}\right)-\mathrm{C}\left(1^{\prime}\right)-\mathrm{N}(1)-\mathrm{C}(6)$ | $(X=\mathrm{H})$ | $(X=\mathrm{P})$ |
| $\chi$ | $\mathrm{C}(8)-\mathrm{C}(7)-\mathrm{C}(5)-\mathrm{C}(6)$ | 64.4 | $79 \cdot 1$ |
|  |  | $12 \cdot 8$ | - |


ig. 2. Projections of a hypothetic adenylyl-( $\left.3^{\prime} \rightarrow 5^{\prime}\right)$-5-isopropyl-2' deoxyuridynyl- $\left(3^{\prime} \rightarrow 5^{\prime}\right)$-adenosine segment of a single strand of $B$-DNA. The pyrimidine base of the title compound was fitted by least squares to that of the $B$-DNA model computed from the coordinates of Arnott, Dover \& Wonacott (1969). H atoms of the terminal methyl groups are drawn with their van der Waals radii. Projection (a) is viewed along the $\mathrm{C}(2) \cdots \mathrm{C}(5)$ direction of the central base, while projection (b) is drawn along the sugarphosphate backbone.
line state may explain the limited incorporation of the corresponding 5 'triphosphate ester of the title compound in enzymatic DNA synthesis (Žmudzka, Nowak \& Sági, 1979).
(c) The overall conformation. The title compound has an anti, gauche-gauche ( $g^{+}$) C( $2^{\prime}$ )-endo envelope conformation. It resembles that in sodium uridine-5' -O -methylphosphate-methanol (Hoogendorp \& Romers, 1978) (Table 4).

Table 5. Hydrogen-bond parameters of the title compound

| $A \cdots \mathrm{H}-D$ | $A \cdots D$ | $A \cdots \mathrm{H}$ | $A \cdots \mathrm{H}-D$ |
| :---: | :---: | :---: | :---: |
| $\mathrm{O}(4) \cdots \mathrm{H}\left(\mathrm{O} 5^{\prime}\right)-\mathrm{O}\left(5^{\prime}\right)^{\mathrm{i}}$ | $2.789(7) \AA$ | $1.887(7) \dot{\mathrm{A}}$ | $173(5)^{\circ}$ |
| $\mathrm{O}(4) \cdots \mathrm{H}\left(\mathrm{O} 3^{\prime}\right)-\mathrm{O}\left(3^{\prime}\right)^{\mathrm{i}}$ | $2.917(8)$ | $2.074(7)$ | $155(6)$ |
| $\mathrm{O}\left(3^{\prime}\right) \cdots \mathrm{H}(\mathrm{N} 3)-\mathrm{N}(3)^{\circ i i n}$ | $2.838(9)$ | $1.949(8)$ | $157(5)$ |

Symmetry code (unlabelled atoms refer to the original molecule)
(i) $1-x, \frac{1}{2}+y, 1-z$
(iii) $2-x, \frac{1}{2}+y,-z$
(ii) $2-x, \frac{1}{2}+y, 1-z$

## Hydrogen bonding

The non-bonded contacts less than the sum of the van der Waals radii indicate three hydrogen bonds in which all the H atoms linked to O and N are involved (Table 5). $O\left(3^{\prime}\right)$ takes part in two helices formed around the screw axes $2_{1}\left(1, y, \frac{1}{2}\right)$ and $2_{1}(1, y, 0)$, partly as donor, partly as acceptor. $\mathrm{O}(4)$ is involved in two helices built up around the screw axes $2_{1}\left(1, y, \frac{1}{2}\right)$ and $2_{1}\left(\frac{1}{2}, y, \frac{1}{2}\right)$ as acceptor.

We thank Mrs Mariann Strumpel (Ann Arbor, Michigan) for providing the least-squares program $A B C L S$.

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[^0]:    * Lists of structure factors and anisotropic thermal parameters have been deposited with the British Library Lending Division as Supplementary Publication No. SUP 34327 ( 7 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CHl 2HU, England.

